

${}_{60}S_8$ and $C_{60}S_{16}$ were reported. It was determined that the interactions between the C_{60} and sulfur consisted mainly of weak van der Waals interactions—a structure in which the C_{60} cages were surrounded by unbound crown-shaped sulfur rings (S_8)^[2-3].

However, in high-pressure Raman studies of $C_{60}S_{16}$ ^[4], it has been proposed that the crown S_8 rings break down and form covalent C-S bonds with the C_{60} cage. The covalent bonding of sulfur directly to C_{60} has been reported in the thermolysis preparation of $C_{120}OS$ ^[5]. A plausible synthesis for $C_{60}S$, however, has yet to be determined^[6]. While $C_{60}S$ has yet to be synthesized, computational results for the [6,6] $C_{60}S$ and [5,6] $C_{60}S$ bridged systems have been

reported at the semi-empirical MNDO ^[6] and AM1 ^[7] levels. A mechanism for the rearrangch i-empirica

Table 1: This table is a summary of the calculated molecular energies for each of the compounds at different levels of theory. Only the initial singlet compounds were treated at the HF/DH(d)+S(d,p) level. Triplet states, protonated compounds and ionized compounds were built from B3LYP/DH(d)+S(d,p) geometries.

The geometry of the [6,6]C₆₀S supports it as the most stable neutral isomer. The crucial C-C bond over which the sulfur atom is bridged was calculated to be 1.541 Å, while in the [5,6] isomer this C-C bond was 1.592 Å, as shown in Table 2. Since the difference in the le

[6,6]C₆₀S, however, is 38.48 kcal mol⁻¹ higher in energy than the singlet state. Therefore, the singlet state is most likely preferred.

Table 2: Bond Distances and Angles at B3LYP/DH(d) + S(d,p) Fine Grid theory level

Compound	Bond	Distance (Å)	Bond	Angle (°)
C ₆₀ S ²⁻	C-C	1.393 - 1.529	S-C-C (to Hexagon)	116.2
	C-S	1.865	S-C-C (to Pentagon)	109.5
[5,6]C ₆₀ S	C-C	1.387 - 1.592	C-S-C	51.1
	C-C (bound to S)	1.592		
	C-S	1.846		
[6,6]C ₆₀ S	C-C	1.387 - 1.541	C-S-C	49.6
	C-C (bound to S)	1.541		
	C-S	1.836		
[5,6]C ₆₀ S ¹⁻	C-C	In Progress	C-S-C	In Progress
	C-C (bound to S)	In Progress		
	C-S	In Progress		
[6,6]C ₆₀ S ¹⁻	C-C	1.390 - 1.532	C-S-C	48.9
	C-C (bound to S)	1.532		
	C-S	1.850		
C ₆₀ SH ¹⁻ (on hexagon)	C-C	In Progress	S-C-C (to Hexagon)	In Progress
	C-S	In Progress	S-C-C (to Pentagon)	In Progress
	S-H	In Progress	C-S-H	In Progress
C ₆₀ SH ¹⁻ (in pentagon)	C-C	In Progress	S-C-C (to Hexagon)	In Progress
	C-S	In Progress	S-C-C (to Pentagon)	In Progress
	S-H	In Progress	C-S-H	In Progress
C ₆₀ S ²⁻ Triplet	C-C	1.396 - 1.531	S-C-C (to Hexagon)	99.8
	C-S	1.863	S-C-C (to Pentagon)	116.8
[5,6]C ₆₀ S Triplet	C-C	1.387 - 1.536	C-S-C	48.6
	C-C (bound to S)	1.536		
	C-S	1.867		
[6,6]C ₆₀ S Triplet	C-C	1.395 - 1.530	C-S-C	49.2
	C-C (bound to S)	1.530		
	C-S	1.838		

Table 2: Unique bond lengths and angles for each of the three studied isomers of C₆₀S.

It is also worthwhile to

It should be noted that these calculations were performed using the energies as calculated at the coarse grid B3LYP/DH(d)+S(d,p) level—fine grid energy is still being calculated for [5,6]C₆₀S.

Table 4: Vertical Electron Affinities (Coarse Grid)

Compound	Electron Affinity (eV)
[5,6]C ₆₀ S	3.797
[6,6]C ₆₀ S	2.477
C ₆₀	2.683 ^[11]
Br ⁻	3.363 ^[12]
I ⁻	3.063 ^[12]
S ⁻	2.07 ^[12]

Table 6: Proto

proton affinity for each of the closed isomers as well as calculating the frequency data for all the aforementioned forms of $C_{60}S$.

While this study brushes the surface of the characterization of $C_{60}S$ at a decent theory level, there is still a need to computationally explore, at an electron correlated theory level, a plausible mechanism of isomer rearrangement as well

[6] D

[6,6]-closed C_{60} S B3LYP/DH(d)+S(d,p)

C	0.0076383625	-0.7015876108	-3.4782106185
C	0.0076383625	-0.7015876108	3.4782106185
C	0.0202950203	0.6964112348	-3.4660658130
C	0.0202950203	0.6964112348	3.4660658130
C	-1.1781221259	-1.4228565406	-3.0338979572
C	-1.1781221259	-1.4228565406	3.0338979572
C	1.1801460427	-1.4442050632	-3.0338983597
C	1.1801460427	-1.4442050632	3.0338983597
C	-1.1522623000	1.4349459211	-3.0202553182
C	-1.1522623000	1.4349459211	3.0202553182
C	1.2060316368	1.4135976284	-3.0202567522
C	1.2060316368	1.4135976284	3.0202567522
C	-2.3073323269	-0.7160536141	-2.6046290159
C	-2.3073323269	-0.7160536141	2.6046290159
C	2.3219676950	-0.7579625855	-2.6046296631
C	2.3219676950	-0.7579625855	2.6046296631
C	-2.2966068097	0.7390620473	-2.6012235463
C	-2.2966068097	0.7390620473	2.6012235463
C	2.3375900965	0.6971087530	-2.6012241559
C	2.3375900965	0.6971087530	2.6012241559
C	-0.7386787490	-2.6033070947	-2.3091512963
C	-0.7386787490	-2.6033070947	2.3091512963
C	0.7193994622	-2.6165006422	-2.3091473201
C	0.7193994622	-2.6165006422	2.3091473201
C	-0.7017424744	2.6091503566	-2.2984941107
C	-0.7017424744	2.6091503566	2.2984941107
C	0.7768452044	2.5957666703	-2.2984958599
C	0.7768452044	2.5957666703	2.2984958599
C	-3.0407791727	-1.1639924336	-1.4284221165
C	-3.0407791727	-1.1639924336	1.4284221165
C	3.0471845750	-1.2191081953	-1.4284234227
C	3.0471845750	-1.2191081953	1.4284234227
C	-3.0158128351	1.1888744680	-1.4287956266
C	-3.0158128351	1.1888744680	1.4287956266
C	3.0648217143	1.1338248826	-1.4287956274
C	3.0648217143	1.1338248826	1.4287956274
C	-1.4423042514	-3.0336547323	-1.1777926644
C	-1.4423042514	-3.0336547323	1.1777926644
C	1.4151186888	-3.0595165961	-1.1777902951
C	1.4151186888	-3.0595165961	1.1777902951
C	-1.4048253342	3.0491960545	-1.1868513571
C	-1.4048253342	3.0491960545	1.1868513571
C	1.4877796810	3.0230	(+)

[6,6]-closed $C_{60}S^{-1}$ B3LYP/DH(d)+S(d,p)

C	0.0075517316	-0.7111596146	-3.4891950699
C	0.0075517316	-0.7111596146	3.4891950699
C	0.0203649695	0.7041773394	-3.4783868314
C	0.0203649695	0.7041773394	3.4783868314
C	-1.1709844305	-1.4235134702	-3.0346485930
C	-1.1709844305	-1.4235134702	3.0346485930
C	1.1729968387	-1.4	